

# Dose Reconstruction of Di(2-ethylhexyl) Phthalate Using a Simple Pharmacokinetic Model

Matthew Lorber, Antonia M. Calafat

http://dx.doi.org/10.1289/ehp.1205182

Online 24 September 2012



National Institutes of Health
U.S. Department of Health and Human Services

Page 1 of 29

Dose Reconstruction of Di(2-ethylhexyl) Phthalate Using a Simple Pharmacokinetic Model

Matthew Lorber, <sup>1</sup> Antonia M. Calafat<sup>2</sup>

<sup>1</sup>Office of Research and Development, U.S. Environmental Protection Agency, Washington, DC, USA; <sup>2</sup>National Center for Environmental Health, Centers for Disease Control and Prevention, Atlanta, Georgia, USA

Address correspondence to M. Lorber, 1200 Pennsylvania Ave. NW, Washington, DC, 20460 USA. Telephone: (703) 347-8535. Fax: (703) 347-8692. E-mail: <a href="mailto:lorber.matthew@epa.gov">lorber.matthew@epa.gov</a>

**Short Running Title:** Dose reconstruction of DEHP

**Key words:** DEHP, dose reconstruction, pharmacokinetic model, phthalate exposure.

**Acknowledgments:** We thank M. Silva, E. Samandar, and J. Preau (CDC) for their assistance in measuring the urinary concentrations of DEHP metabolites; and A. Sjodin for designing the original study to assess the variability of urinary concentrations of polycylic aromatic hydrocarbon metabolites and for providing the samples that we used for the present analysis. We

also thank Paul Lioy (EOHSI, Robert Wood Johnson Medical School and Rutgers University) for constructive discussions during manuscript development.

**Disclaimer:** The findings and conclusions in this report are those of the authors and do not necessarily represent the views of the Environmental Protection Agency or the Centers for Disease Control and Prevention.

**Financial Interests Declaration:** The authors declare they have no actual or potential competing financial interests.

### **Abbreviations**

DEHP di(2-ethylhexyl) phthalate

MECPP mono(2-ethyl-5-carboxypentyl) phthalate

MEHHP mono(2-ethyl-5-hydroxyhexyl) phthalate

MEHP mono(2-ethylhexyl) phthalate

MEOHP mono(2-ethyl-5-oxohexyl) phthalate

NHANES National Health and Nutrition Examination Survey

PAH polycyclic aromatic hydrocarbon

RfD reference dose

## **ABSTRACT**

**Background:** Di(2-ethylhexyl) phthalate (DEHP), used primarily as a plasticizer for polyvinyl chloride, is found in a variety of products. Previous studies have quantified human exposure by back-calculating intakes based on DEHP metabolite concentrations in urine, and by determining concentrations of DEHP in exposure media (e.g., air, food, dust).

**Objectives:** To better understand the timing and extent of DEHP exposure, we used a simple pharmacokinetic model to "reconstruct" the DEHP dose responsible for the presence of DEHP metabolites in urine.

**Methods:** We analyzed urine samples from eight adults for four DEHP metabolites [mono(2-ethylhexyl) phthalate, mono(2-ethyl-5-hydroxyhexyl) phthalate, mono(2-ethyl-5-oxohexyl) phthalate, and mono(2-ethyl-5-carboxypentyl) phthalate]. Participants provided full volumes of all voids over one week and recorded the time of each void and information on diet, driving, and outdoor activities. Using a model previously calibrated on a single person self-dosed with DEHP in conjunction with the eight participants' data, we used a simple trial-and-error method to determine times and doses of DEHP that resulted in a best fit of predicted and observed urinary concentrations of the metabolites.

**Results:** The average daily mean and median reconstructed DEHP doses were 10.9 and 5.0 μg/kg-day, respectively. The highest single modeled dose of 60 μg/kg occurred when one study participant reported consuming coffee and a bagel with egg and sausage that was purchased at a gas station. About two-thirds of all modeled intake events occurred near the time of reported

food or beverage consumption. Twenty percent of the modeled DEHP exposure occurred between 2200 hours and 0500 hours.

**Conclusions:** Dose reconstruction using pharmacokinetic models—in conjunction with biomonitoring data, diary information, and other related data—can provide a powerful means to define timing, magnitude, and possible sources of exposure to a given contaminant.

#### INTRODUCTION

Phthalates, diesters of phthalic acid, are a class of industrial chemicals extensively used as softeners of plastics, solvents in perfumes, and additives to many personal care and consumer products such as hairsprays, lubricants, and insect repellents (David et al. 2001; Koch and Calafat 2009). Di(2-ethylhexyl) phthalate (DEHP) is used primarily as a plasticizer for polyvinyl chloride and can be found in a variety of products, such as floor or wall coverings, vinyl gloves, toys and child care articles, materials that have contact with foods, and medical devices (Schettler 2006).

Several phthalates, including DEHP, di-n-butyl phthalate, diisobutyl phthalate, and butyl benzyl phthalate, have been identified as endocrine disruptors in animal studies (National Research Council 2008). In the body, phthalates rapidly hydrolyze to their respective monoesters; some monoesters are further metabolized by phase I and/or phase II reactions. For phthalates with short alkyl side chains, monoesters represent the major human metabolites. By contrast, for phthalates with longer alkyl chains, including DEHP, the main metabolites are the products of  $\omega$ -,  $\omega$ -1, and  $\beta$ -oxidations of the alkyl chain (Koch and Calafat 2009). All phthalate metabolites are excreted in the urine or feces within a few hours; excretion is complete within 1 or 2 days (Koch and Calafat 2009).

Two approaches have been used to quantify human exposures to DEHP and other phthalates. One relies on the biomonitoring of DEHP metabolites in urine to back calculate the daily intake of DEHP. For example, researchers have used biomonitoring data from the United States (Kohn et al. 2000; Lorber et al. 2010), Germany (Koch et al. 2003; Wittassek et al. 2007), and other countries (Fujimaki et al. 2006; Huang et al. 2006) to characterize general population

exposures. The other approach uses DEHP concentrations in exposure media (e.g., air, food, dust) and exposure contact rates to estimate daily intakes. This approach has been used in two studies, one using worldwide exposure media data (Clark et al. 2011) and the other using European exposure data (Wormuth et al. 2006). Both approaches arrived at similar intakes of DEHP in adults, ranging from about 2 to 10 µg/kg-day. These researchers concluded that, for the general population, diet explained the bulk of exposure to DEHP.

For the present study, we used a third approach to assess general population exposures to DEHP. This approach relies on having precise times of urination and urine volumes, along with urinary concentrations of DEHP metabolites. We used a calibrated simple pharmacokinetic model for DEHP (Lorber et al. 2010) in conjunction with these data to "reconstruct" the dose necessary to have resulted in the observed metabolite concentrations.

### **METHODS**

Biomonitoring data set. We used data collected from four adult men and four adult women, 24–59 years of age, who provided the full volume and time of all urinary void events for 1 week in October–November 2005. Each participant provided diary information that included when and what the participant ate and drank, time spent driving and putting gasoline in the car, as well as time spent in other activities that might influence the appearance of DEHP metabolites and other contaminants in urine. Because we had no information regarding the participants' body weights, we assumed body weights of 70 kg for men and 60 kg for women. The study design has been described in detail previously (Li et al. 2010; Preau et al. 2010; Ye et al. 2011). The institutional

review board of the Centers for Disease Control and Prevention approved the original study; the present study was exempted, and all participants provided written informed consent at the time of the original study.

The data set comprised 56 person-days of data (8 people × 7 days) and included 427 distinct urine samples. These samples have previously been analyzed for several polycyclic aromatic hydrocarbon (PAH) metabolites (Li et al. 2010), bisphenol A (Ye et al. 2011), and monoethyl phthalate (a metabolite of diethyl phthalate) and two DEHP metabolites [mono(2-ethylhexyl) phthalate (MEHP) and mono(2-ethyl-5-hydroxyhexyl) phthalate (MEHHP)] (Preau et al. 2010). For the present study, we used the urinary concentrations of MEHP and MEHHP (Preau et al. 2010), as well as two other DEHP metabolites [mono(2-ethyl-5-oxohexyl) phthalate (MEOHP) and mono(2-ethyl-5-carboxypentyl) phthalate (MECPP)] that had been measured at the time of the original study but not reported by Preau et al. (2010). We also used times of urination, volume of urine voids, and diary information for each person.

**Dose reconstruction**. We used a DEHP model described previously (Lorber et al. 2010). This model is a simple two-compartment empirical model that was structured and calibrated based on the experimental data presented by Koch et al. (2005). The calibration (Lorber et al. 2010) provided all required model rate constants and other parameters. For the present study, the only required inputs were time and volume of each urine void. The dose reconstruction exercise entailed determining the time and extent of DEHP exposure that would lead to predicted concentrations matching the measured concentrations of DEHP metabolites. Details on the structure of the model and its prior calibration are provided in Supplemental Material (pp. 3–7).

To reconstruct the DEHP dose, we observed when urinary concentrations of DEHP metabolites increased and then assumed that this increase was related to a recent exposure event. By trial and error, we calibrated the extent and time of exposure to fit the observed urinary concentrations. The calibration of the model entailed visual inspection of the predicted concentrations compared with the observed concentrations. For example, if the MEHP concentration increased from approximately  $10~\mu g/L$  to  $> 200~\mu g/L$  from one urination to the next (and the other metabolites also showed meaningful increases), we assumed that the rise was due to an exposure event that occurred after the first urination event but before the second one. By choosing different times of intake (i.e., exposure event times) and different intake quantities, we found an intake–time pair that resulted in what appeared to be the best fit of predicted and observed concentrations for all four metabolites; however, this fit may not have been optimal for all metabolites. A key assumption of the model is that DEHP exposure occurred as a bolus dose during a single time step of the model, which is consistent with an exposure via food consumption.

Three sensitivity analyses tests were performed. In one of these tests, we assumed that the DEHP dose was spread out over hours (i.e., 2, 8, and 24 hr) in 15-min time intervals around the calibrated time of exposure, rather than assuming exposure to a single bolus dose (for details see Supplemental Material, pp. 7–8). Specifically, we set half of the spread time before and half after the calibrated time of exposure. For the 24-hr simulations, we summed the total intakes for the day and assumed that exposure was evenly spread throughout the day; for the other two simulations, individual exposures were only spread out for the time periods evaluated (2 and 8 hr). In a second sensitivity test, we investigated timing of exposure by modeling a bolus dose occurring 2 hr before or 2 hr after the calibrated time of exposure. Last, we examined the final

reconstructed daily dose and compared to it to daily doses that might be determined from applying the creatinine correction approach to individual spot samples (for details see Supplemental Material, pp. 9–12).

### **RESULTS**

Best fit to the data. Using the rate constants from Lorber et al. (2010) for the eight study participants, we found a good fit to the data. This suggested validity of the calibrated parameters, which were determined experimentally based on one individual and then applied to the eight participants. However, a pattern observed in the original experimental data set used to calibrate the model (Koch et al. 2005) suggested a refinement to the best-fit solution in the current model application. Specifically, after approximately 24 hr, a small portion of the original dose of DEHP appeared to metabolize anew, in almost a second phase of metabolism. That is, the excretion of DEHP metabolites rose slightly on the second day in contrast to a continued first-order decline and loss of the metabolites (Lorber et al. 2010). We hypothesized that some of the spikes in urinary DEHP metabolite concentrations observed in our study population may have reflected a second phase of metabolism from earlier exposures rather than a unique new exposure, particularly when the metabolite concentrations were low and the calibrated dose was also small (near 1 µg/kg) (see Supplemental Material, pp. 5–7 for additional discussion). Therefore, we removed small, calibrated exposures in a series of step-wise exclusions (Table 1). Removing modeled exposures < 1 µg/kg reduced the number of modeled intakes from 111 to 96 and increased the number of days with no exposures from 1 to 4, but had virtually no effect on the measures of central tendency (i.e., mean, median) compared to estimates from the simulation

with all intakes. However, when all exposures < 2  $\mu$ g/kg were removed, the number of days with no exposures rose to 14 and the median intake declined from 5.0 to 4.5  $\mu$ g/kg-day. We observed a more significant impact when we removed all events resulting in exposures < 3  $\mu$ g/kg. Based on these results, we removed all events with < 1  $\mu$ g/kg for the final calibration. Additional details and analyses of secondary metabolism are provided in Supplemental Material, pp. 5–7 and Table S1.

Characteristics of the best fit. Figure 1 shows comparisons between predicted and observed urine concentrations of the four metabolites over a 7-day period for one participant (Subject 1; see Supplemental Material Figures S3 – S9 for Subjects 2–8). The model had a tendency to overpredict the urinary concentrations of MEHP, the hydrolytic monoester metabolite of DEHP (Figure 1A). MEHP has the most rapid elimination in humans, with a 5 hr half-life, and MEHP also represents the smallest amount of the original DEHP dose (5.9%) (Koch et al. 2005). Therefore, because the secondary metabolites of DEHP are considered better indicators of exposure than MEHP (Barr et al. 2003), we chose to focus on finding the model with the best fit for the concentrations of MECPP, MEOHP and MEHHP, at the expense of those of MEHP.

Although a visual inspection of Figure 1 suggests that trends in metabolite concentrations were well captured by the model, this may be at least in part a visual artifact. For example, the model seems to have predicted the observed high MEHHP concentration on day 3 (Figure 1B). However, the observed concentration of MEHHP for the next observation is 195 µg/L, whereas the model predicted 488 µg/L. For MECPP, the model predictions and observations are consistent for this time point, with both values near 400 µg/L (Figure 1C), but for MEOHP, the

observed value is higher than the prediction (320  $\mu$ g/L versus 112  $\mu$ g/L, respectively; Figure 1D). This illustrates the trade-offs resulting from the use of the trial-and-error approach to find the "best fit" reconstructed dose for each individual over 7 days.

Figure 2 shows predicted versus observed concentrations of the four metabolites for all eight individuals and their 427 urination events. The correlations were excellent, with correlation coefficients (r) between 0.85 and 0.90. However, for MEHP, the model tended to overpredict concentrations (Figure 2A). Specifically, the average predicted concentration (25 µg/L) is just over twice the average observed concentration (12 µg/L). In contrast, the average predicted and observed concentrations of MEHHP (Figure 2B) are virtually identical (88 and 87 µg/L, respectively), with a similar spread of values above and below the best-fit line. The model has a slight tendency to underpredict MECPP concentrations, with an average predicted value of 79 µg/L, compared with an average observed value of 99 µg/L (Figure 2C). Results for MEOHP are similar to those of MEHHP – predictions and observations are virtually identical with both averaging 53 µg/L with a high correlation coefficient of 0.87.

Characteristics of Exposure. The best-fit solution (baseline minus events with exposures <  $1\mu g/kg$  body weight; Table 1) contained 96 distinct exposure events, or about 1.7 events/person-day (96 events  $\div$  56 person-days). Most person-days included 0, 1, or 2 exposure events per day, but there were 6 days with 3 events and 2 days with 4 events. The modeled mean and median daily DEHP exposures were 10.9 and 5.0  $\mu g/kg$ -day, respectively. The daily mean intakes for the eight participants, from lowest to highest were: 3.3, 4.3, 5.0, 9.8, 13.4, 15.4, 17.7, and 18.3  $\mu g/kg$ -day. The distribution in selected exposure ranges for the 56 person-days of daily

intakes is presented in Table 2. The highest single person-day of exposure was 80  $\mu$ g/kg-day, with the second highest being 36  $\mu$ g/kg-day.

Figure 3 shows the estimated average hourly DEHP intakes for the eight participants. The higher hourly averages for 0800 to 1700 hours suggest that about 60% of the DEHP intakes occurred when participants were away from home, although the study period did include 2 weekend days where the exposures were more likely to occur at home.

We also superimposed the modeled time of exposure on participants' diary information (data not shown). Of the 96 exposure events, 66 were predicted by the model to have occurred between 0500 and 2200 hours; of these, 42 were predicted to have occurred within 1 hr (before or after) of reported consumption of food and/or beverage. The single highest modeled exposure event, at 60 µg/kg, occurred at the same time that the participant reported consuming "bagel w/egg, sausage, cheese, coffee bought at gas station." We speculate that these food items might have been kept warm and stored in a setting and/or in contact with materials that contained and transferred DEHP to the food.

Of 96 exposure events, 30 occurred at night (between 2200 and 0500 hours), and the average nighttime exposure was 2.2  $\mu$ g/kg-day, or about 20% of the daily total. None of the study participants reported food or beverage consumption at night during the study. However, model estimates indicated that one person with a relatively low average exposure ( $\sim$  3  $\mu$ g/kg-day) had only nighttime exposures. In addition, the person with the highest average daily modeled DEHP exposure (18.3  $\mu$ g/kg-day) experienced 41% of his/her exposure ( $\sim$  7.5  $\mu$ g/kg-day on average) at night. Figure 4 shows examples of two nighttime exposure events and one daytime exposure event from this person. No reported activities in the person's diary coincided

with the nighttime exposure episodes at 2330 hours (20  $\mu$ g/kg) and 0245 hours (10  $\mu$ g/kg). The daytime exposure of 32  $\mu$ g/kg, the second highest modeled exposure event in this study, occurred at 1000 hours on the seventh day of the experiment. No food or beverage consumption events were reported from 0900 to 1100 hours on that day, but the participant did report consumption of cereal at 0600 hours and tea at 1130 hours.

**Sensitivity Analyses.** We ran additional simulations (see Supplemental Material, pp. 7– 8) that assumed the DEHP exposure events were spread out around the calibrated time point, in time periods of 2, 8, and 24 hr instead of 15 min. With 2-hr exposure events, the results were virtually identical (see Supplemental Material, Figure S1 for example data for subject 3). With exposure episodes of 8 hr, we observed a slight change in the location of the spikes in urinary concentrations of DEHP metabolites, but the trends remained similar. In contrast, for 24-hr exposure events, the predicted concentrations were discernibly different than the actual observations. The fact that the 2-hr spread resulted in virtually identical concentration predictions is expected from a modeling perspective, and possibly in reality. The model calculated void concentrations based on the mass of metabolite in the bladder at the time of the void event. The same mass would arrive in the bladder whether the modeled exposure was from small bolus doses spread out over a period of time prior to the void event or from one large bolus dose at a single time before the void event. However, as the large dose is spread into smaller and smaller bolus doses over a longer period of time, doses begin to overlap void events so that different amounts are delivered and available for each void, compared with the single large bolus scenario.

We conducted an additional sensitivity analysis to compare our model-based dose reconstruction with DEHP intakes calculated from creatinine-corrected metabolite concentrations in single spot urine samples, as described in Supplemental Material (pp. 9–12). Specifically, we estimated intakes based on the 4 individual metabolites and their sum for 62 individual urine samples collected from one study participant over the week of sampling, and for 42 individual urine samples collected between 0700 and 1900 hours each day for that study participant. The average model-based daily intake for this participant (18.3 µg/kg-day) was similar to the average daily intake as calculated by the creatinine correction approach (19.4 µg/kg-day) (Supplemental Material, Table S2). However, daily predictions of intakes based on a single urine void using the creatinine correction approach showed wide variability. For example, the average creatininepredicted daily intake from 9 void events on one day was 46.9 µg/kg-day based on the sum of the four metabolites, compared with the model-based reconstructed dose of 36.0 µg/kg-day (Thursday, Supplemental Material, Table S2). The range in predictions over these 9 events was very wide, however, at between 4.0 and 99.5 µg/kg-day. We observed similar wide ranges in all days of the week, with the difference in high and low predictions ranging from a factor of 4 to a factor of 25. Further, if one considered only the void events during daytime hours [when an individual might logically contribute a sample to a survey program such as the National Health and Nutrition Examination Survey (NHANES)], the average predicted intake based on creatinine correction declined to 16.4 µg/kg-day (Supplemental Material, Table S3). This finding of a difference between daytime and nighttime extrapolated intakes is similar to what Preau et al. (2010) found for MEHHP metabolite concentrations in this cohort overall; they found that the geometric mean concentration of samples collected in the evening, 33.2 µg/L, was significantly higher (p < 0.01) than in samples collected in the morning, 18.7  $\mu$ g/L, or in the afternoon, 18.1

μg/L. At least for this individual and this cohort, urine samples taken during the day and used to extrapolate intakes using the creatinine correction approach would result in an underestimate of overall intakes.

### **DISCUSSION**

The model used in the present study appeared to accurately predict DEHP metabolite urine concentrations in eight adult volunteers who provided urine samples over a 7-day period, and our estimates of daily total exposures to DEHP were consistent with previous exposure estimates determined through other means for other populations. For example, the average DEHP intake measured in full-day diet samples consumed by a group of German adults was 2.4 µg/kg-day (Fromme et al. 2007). Clark et al. (2011) estimated median intakes of DEHP of 11.0 µg/kg-day, using worldwide data on concentrations of DEHP in exposure media in combination with contact rates. Wormuth et al. (2006) conducted a similar analysis, and their "intermediate" estimate of intake characterized as specific to Europe (although like Clark, they used concentration data from around the world) was 2.1 µg/kg-day. Both Wormuth and Clark concluded that food pathways dominated for DEHP. Daily DEHP exposures have also been determined using surveys of DEHP metabolites in urine. Wittassek et al. (2007) estimated a median DEHP intake of 3.5 µg/kg-day using 24-hr urine samples from the German Environmental Specimen Bank for Human Tissues, collected from 634 German students between 1988 and 2003. Kohn et al. (2000) estimated a median intake of 0.7 µg/kg-day using MEHP urinary data from NHANES III, corresponding to years 1988–1994. Lorber et al. (2010)

estimated a mean US intake of DEHP in the range of  $0.2 - 2.2 \,\mu\text{g/kg-day}$  based on DEHP metabolites data from NHANES 2001–2002.

Our analysis also suggested that DEHP exposure was dominated by the diet, as reported by previous studies (Clark et al., 2011; Wormuth et al., 2006). However, for some individuals, significant DEHP exposures occurred at night, between the 2200 and 0500 hours, when dietary exposures are unlikely. We used a modeling strategy to reconstruct DEHP exposures as single bolus events; possible nighttime bolus activities that could result in DEHP exposure include, for example, ingestion of water (or other beverage), and use of medications or personal care products. However, if the exposure had occurred over a number of hours instead of as a bolus dose as modeled, other possible exposures should be considered [e.g., inhalation exposures from use of a continuous positive airway pressure machine used for sleep apnea and/or snoring, oral exposure from a plastic mouth guard, emission of DEHP from vinyl materials (e.g., flooring) in the participant's home]. Results of our sensitivity analysis suggested that it is possible that a continuous inhalation exposure while the person was asleep could also explain the observed concentrations. In addition, moving the calibrated bolus intakes 2 h before or 2 h after the calibrated time did make a difference in the predictions, such that a calibrated time of exposure of 0200 hours does suggest, in fact, post-midnight exposure rather than pre-midnight exposure.

The tendency of our best-fit solution to overpredict MEHP might be explained by the use of experimental data from a male in his 60s (Koch et al. 2005) in the calibration of the model (Lorber et al. 2010). All eight participants in the present study were < 60 years of age, and seven of them were < 40 years of age. It could be that the older man excreted more MEHP than the younger persons; his body was not as efficient in quickly metabolizing MEHP as the younger

individuals. Nonetheless, modeled concentrations for MEHP were lowest of the four metabolites. The metabolism and excretion of the other three metabolites were well predicted with no change in calibrated rate parameters. Overall, we consider the model to be valid for this type of application.

About one-third of the person-days (18 of 56) had daily exposures ≥ 50% of the U.S. Environmental Protections Agency's (EPA) reference dose (RfD) for DEHP of 20 µg/kg-day (U.S. EPA 2012), and about 18% of the person-days had exposures above the RfD. The U.S. EPA has defined the RfD as "an estimate (with uncertainty spanning perhaps an order of magnitude) of a daily exposure to the human population (including sensitive subgroups) that is likely to be without an appreciable risk of deleterious effects during a lifetime." (U.S. EPA 2012). While some participants in our study population—which is not representative of any specific population—experienced intakes at or near the RfD as estimated by the dose reconstruction, we cannot draw conclusions regarding the potential for health impact from such exposures to DEHP.

The potential for studying exposure to chemicals ubiquitous in consumer products with our approach is clear. We were fortunate to have a previously calibrated model as well as an experimental cohort with pertinent biomonitoring information (measurements in all collected urine samples) and diary information. Understandably, availability of this type of model and experimental data are rare. With our model and data, we reconstructed the exposures of 8 individuals to DEHP over 7 days, albeit with these provisos: 1) very small perturbations in urinary concentrations might not signal a unique exposure, but rather a second phase of metabolism from an earlier exposure, 2) the data on which the model were calibrated showed

higher urinary concentrations of MEHP than the experimental cohort (speculated to be the result of age in the individual whose data was used to originally calibrate the model), and 3) it is possible that the exposure event which caused a rise in DEHP metabolites concentrations in urine occurred over a period of time before the urinary event rather than as a bolus event.

Since the publication of the calibrated model used in this study, an extensive data set on human toxicokinetics of DEHP has become available (Anderson et al., 2011). This data set included twenty volunteers dosed with stable-isotope labeled DEHP and their urine sampled and analyzed for DEHP metabolites over 48 hours, similar to the study on the single individual (Koch et al., 2005) used to calibrate this study's model. Anderson et al. (2011) compared their data to Koch's data, and noted a similarly rapid elimination of metabolites with most excreted within 24 hours. However, Anderson et al. (2011) also noted that the four key metabolites of DEHP explained less of the dose compared with Koch's experiment (i.e., the molar fractions of excretion they derived for each DEHP metabolite were lower than the molar fractions for each metabolite derived from Koch's data). They discuss the implication of this difference in extrapolating results from a single spot urine sample to a daily exposure—that difference being that their extrapolation factors would lead to higher daily estimated intakes as compared to extrapolation factors based on Koch's data. We would expect a similar impact if the model we used was recalibrated on Anderson's data—that higher daily constructed doses would emerge.

Given these qualifications, some of the results were nonetheless compelling. For example, we believe that the reporting of consumption of gas station food and a relatively high DEHP exposure calibrated to occur around the same time is unlikely to be a coincidence. The finding that certain individuals have meaningful nighttime DEHP exposures warrants follow-up.

While certainly data intensive, a key advantage to the approach described in this paper is that it can reduce variability compared with the widely used creatinine correction approach. A sensitivity analysis compared our reconstructed daily intake estimate with estimates based on creatinine-corrected metabolite concentrations in individual urine samples taken during the day from a single participant. The reconstructed intake was 18.3 µg/kg-day, which was similar to the average intakes surmised by the creatinine correction approach for 9 urine events during the day, at 19.4 µg/kg-day. However, the range of creatinine-calculated intakes from the 9 events was about 4 to 100 µg/kg-day. At least for this individual, overall intakes would be underestimated if based on the creatinine correction approach.

Although it is data intensive, the modeling approach advocated in this study has the potential not only to refine estimates of daily intakes and their variability, but to provide insight into potential pathways of exposure. In follow-up studies, a logical strategy would be to first screen a large number of individuals who have provided one urine sample to identify a smaller set of individuals that appear to have relatively high exposures. Then, this smaller set could be tracked over time with additional urine collections and comprehensive diaries to identify sources and pathways of exposure to DEHP. As for all phthalates, data on the DEHP content of consumer use products are limited (Kawakami et al. 2011). Several studies have concluded that DEHP exposure is dominated by food, but although packaging and/or preparation with products containing DEHP has been proposed as the cause of food contamination, this has not been studied to any extent. While characterizing phthalate exposure through urine sampling is now and will likely remain the primary means to study total exposure, studies that use detailed measurement data, diary data, and modeling can identify sources of exposure and possibly shed light on pathways not otherwise considered.

#### REFERENCES

Anderson WAC, Castle L, Hird S, Jeffrey J, Scotter MJ. 2011. A twenty-volunteer study using deuterium labeling to determine the kinetics and fractional excretion of primary and secondary urinary metabolites of di-2-ethylhexylphthalate and di-is-nonylphthalate. Food Chem Tox 49: 2022-2029.

Barr DB, Silva MF, Kato K, Reidy JA, Malek NA, Hurtz D, et al. 2003. Assessing human exposure to phthalates using monoesters and their oxidized metabolites as biomarkers. Env. Health Persp. 111: 1148-1151.

Clark KE, David RM, Guinn R, Kramarz KW, Lampie MA, Staples CA. 2011. Modeling Human Exposure to Phthalate Esters: A Comparison of Indirect and Biomonitoring Estimation Methods. Human and Eco. Risk Assess 17: 923-965.

David RM, McKee RH, Butala JH, Barter RA, Kayser M. 2001. Esters of aromatic mono-, di-, and tricarboxylic acids, aromatic diacids, and di-, tri-, or polyalcohols. In: Patty's Toxicology, Vol 6 (Bingham E, Cohrssen B, Powell CH, eds). 5th ed. New York: John Wiley and Sons, 635–932.

Fromme H, Gruber L, Schlummer M, Wolz G, Bohmer S, Angerer J, et al. 2007. Intake of phthalates and di(2-ethylhexyl)adipate: Results of the Integrated Exposure Assessment Survey based on duplicate diet samples and biomonitoring data. Environ Int 33:1012–20.

Fujimaki K, Yoshinaga J, Watanabe C, Serizawwa S, Shiraishi H, Mizumoto Y. 2006. Estimation of intake level of di (2-ethyhexyl) phthalate (DEHP) in Japanese pregnant women based on measurement of concentrations of three urinary metabolites. Nippon Eiseigaku Zasshi (Japanese Journal of Hygiene) 61:340–7.

Huang PC, Kuo PL, Lee CC. 2006. The association between urinarymonoester phthalates and thyroid hormone in pregnant women. Epidemiology 17(6) Suppl:S195, November 2006.

Kawakami T, Isama K, Matsuoka A. 2011. Analysis of phthalic acid diesters, monoester, and other plasticizers in polyvinyl chloride household products in Japan. J Environ Sci Health A Tox Hazard Subst Environ Eng 46:855-64.

Koch HM, Calafat AM. 2009. Human body burdens of chemicals used in plastic manufacture. Philos Trans R Soc Lond B Biol Sci 364:2063–78.

Koch, HM, Drexler H, Angerer J. 2003. An estimation of the daily intake of di(2-ethylhexyl)phthalate (DEHP) and other phthalates in the general population. J. Hyg Environ. Health 206:77-83.

Koch HM, Bolt HM, Preuss R, Angerer J. 2005. New metabolites of di(2-3thylhexyl)phthalate (DEHP) in human urine and serum after single oral doses of deuterium-labelled DEHP. Arch Toxicol 79:367-76.

Kohn MC, Parham F, Masten SA, Portier CJ, Shelby MD, Brock JW, et al. 2000. Human exposure estimates for phthalates. Env Health Persp 108: A440-A442.

Li Z, Romanoff LC, Lewin MD, Porter EN, Trinidad DA, Needham LL, et al. 2010. Variability of urinary concentrations of polycyclic aromatic hydrocarbon metabolite in general population and comparison of spot, first-morning, and 24-h void sampling. J Expo Sci Env Epi 20:526–35.

Lorber M, Angerer J, Koch HM. 2010. A simple pharmacokinetic model to characterize exposure of Americans to Di-2-ethylhexyl phthalate. J Expo Sci Env Epi 20:38-53.

National Research Council. 2008 Phthalates and Cumulative Risk Assessment: The Task Ahead. Washington, DC:National Academies Press.

Preau JL, Wong LY, Silva MJ, Needham LL, Calafat AM. 2010. Variability Over 1 Week in the Urinary Concentrations of Metabolites of Diethyl Phthalate and Di(2-Ethylhexyl) Phthalate Among Eight Adults: An Observational Study. Env Health Persp. 118:1748-54.

Schettler T. 2006. Human exposure to phthalates via consumer products. Int J Androl 29:134–9.

U.S. EPA (U.S. Environmental Protections Agency). 2012. Di(2-ethylhexyl)phthalate (DEHP) (CASRN 117-81-7). Available: http://www.epa.gov/iris/subst/0014.htm [accessed 24 August 2012].

Wittassek M, Wiesmüller GA, Koch HM, Eckard R, Dobler L, Müller J, et al. 2007. Internal phthalate exposure over the last two decades - a retrospective human biomonitoring study. Int J Hyg Environ Health 210:319-333.

Wormuth M, M Scheringer, M Vollenweider, K Hungerbuhler. 2006. What are the sources of exposure to eight frequently used phthalic acid esters in Europeans? Risk Analysis 26(3):803-824.

Xu Y, Cohen Hubal EA, Little JC. 2010. Predicting residential exposure to phthalate plasticizer emitted from vinyl flooring: sensitivity, uncertainty, and implications for biomonitoring. Environ Health Perspect 118:253-8.

Ye X, Wong L-Y, Bishop AM, Calafat AM. 2011. Variability of Urinary Concentrations of Bisphenol A in Spot Samples, First Morning Voids, and 24-Hour Collections. Env. Health Persp. 119:983-988.

Table 1. Comparison of model estimates from the initial "baseline" calibration (when all rises in urine DEHP metabolite concentrations were assumed to be the result of a unique exposure) with estimates generated after removing exposure events less than 1, 2, and 3  $\mu$ g/kg DEHP.

Description	Baseline	Baseline minus events less than	Baseline minus events less than	Baseline minus events less than
Number of exposure events	111	1 μg/kg 96	2 μg/kg 71	3 μg/kg 56
Average median daily exposure, µg/kg-day	5.0	5.0	4.5	3.5
Average mean daily exposure, µg/kg-day	11.1	10.9	10.4	9.8
Number of person days with no exposure	1	4	14	21

Table 2. The distribution of exposure ranges for the 56 person-days of daily intakes.

Range (µg/kg-day)	Person-days
0	4
1–5	21
6–10	13
10–20	8
> 20	10

# **Figure Legends**

Figure 1. Observed and predicted urinary concentrations of the four DEHP metabolites, MEHP, MECPP, and MEOHP, for Subject 1 throughout the study period. (See Supplemental Material, Figures S1–S7 for corresponding figures for Subjects 2–8.)

Figure 2. Correlations between observed and predicted urinary concentrations of the four DEHP metabolites, MEHP, MEHPP, and MEOHP, for all 427 urination events for the eight study participants.

Figure 3. Average hourly model-based DEHP intakes for the eight study participants. The dotted lines highlight the intakes that occurred between 0800 and 1700 hrs.

Figure 4. Observed (Obs) and predicted (Pred) urinary concentrations of MEHHP for the individual with highest nighttime DEHP intake (Subject 3). Boxes show the time of day, volume of urine , and urinary concentrations ( $\mu$ g/L) of MEHHP. The time of day and magnitude of the calibrated exposure event are shown in circles.

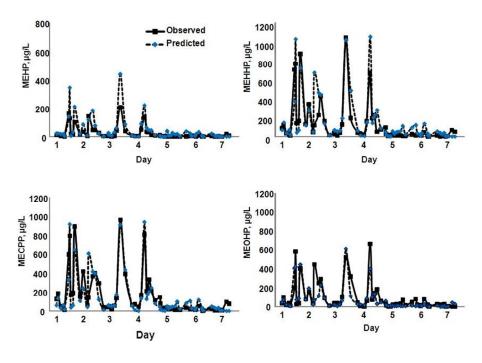


Figure 1 215x140mm (111 x 111 DPI)

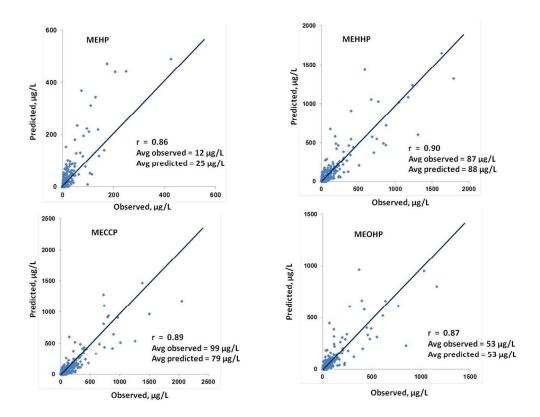


Figure 2 235x180mm (150 x 150 DPI)

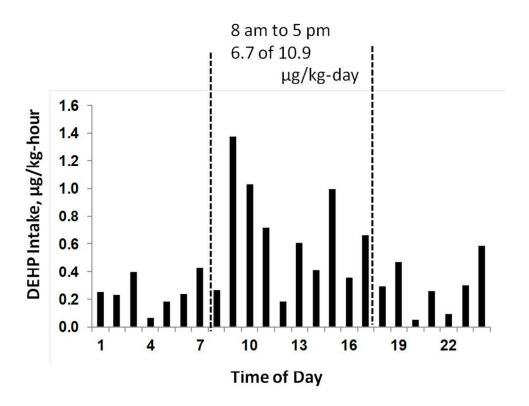


Figure 3 170x157mm (150 x 150 DPI)

7

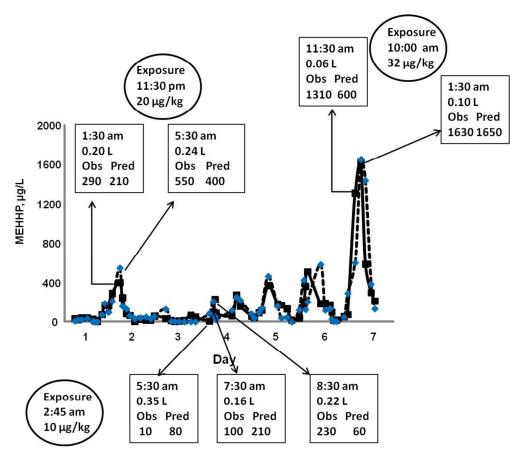


Figure 4 194x168mm (150 x 150 DPI)